BRIDGEHEAD NITROGEN HETEROCYCLICS VIA IMIDE REDUCTION.

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Incorporation of imides in condensed heterocyclics via a Bischler-Napieralski type of ring closure often leads to low yields of cyclization products¹. On the other hand the continuing interest² in the chemistry of the indole and isoquinoline alkaloids emphasizes the potential value which a general method of synthesis for these types of compound might have. As reported recently the acid-catalyzed cyclization³ of 5-hydroxy-2-pyrrolidinones and 5-alkoxy-2-pyrrolidinones offers an attractive method to circumvent the difficulties in working with concentrated inorganic acids or phosphorus oxychloride⁴.

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The hydroxy compounds $\underline{2a-2e}$ were prepared via p_H controlled NaBH₄ reduction of the corresponding cyclic imides $\underline{1a-1e}$ and "base work-up". The corresponding ethoxy derivatives $\underline{3a-3e}$, generally obtained as slightly unstable oils in quantitative yield, could be synthesized following an analogous procedure except that work-up of the compounds was carried out at p_H <7("acid work-up"). Pertinent analytical data on both types of compound are listed in the table.

Cyclization of <u>2b</u>³, <u>2e</u>, <u>3b</u>, <u>3c</u>, <u>3d</u> and <u>3e</u>, in which the aryl nucleus is activated by a methoxy substituent, was carried out in refluxing benzene/
TsOH and proceeded under mild conditions and in high yields to the corresponding ring closed products.

TABL	Ξ

Compound	a mp	b IR	NMR C	yield
<u>2a</u>	120-125°	1660/cm	4.9 - 5.05	75%
<u>2b</u>	107-110°	1640	4.9 - 5.1	70
<u>2c</u> (X=H)	80-82°	1660	5.0 - 5.15	80
<u>2d</u> (X=OMe)	104-110°	1650	5.1 - 5.25	93
<u>2e</u> (X=H)	113 - 115°	1660	4.75 - 4.95	89
<u>3a</u>	-	1685/cm	4.5 - 4.75	
<u>3b</u>	-	1685	4.6 - 4.85	
<u>3c</u> (X=H)	-	1675	4.78 - 4.91	
<u>3c</u> (X=OMe)	-	1675	4.75 - 4.95	
<u>3d</u> (X=OMe)	_	1680	4.85 - 5.05	
<u>3e</u> (X=H)	-	1680	4.4 - 4,6	

a) the sometimes relatively large melting range is due to thermal instability.

b) Solids determined in KBr; oils in CHCl₃.

Thus the $\Delta 9,11-13$ -azasteroid $\underline{4a}$ could be prepared directly via the cyclization of either $\underline{2d}$, $\underline{3d}$ or a mixture of $\underline{2d}$ and $\underline{3d}$ in a yield of 46%, mp. 150-152°, NMR (CDC1 $_3$) δ 6.14 (\underline{H}_{11} , X part of ABX). From the mother liquor the known 1h $\Delta 8,9$ -steroid $\underline{4b}$ was obtained in 33% yield after HC1-MeOH isomerisation of the remaining traces of the $\Delta 9,11$ -isomer $\underline{4a}$ into $\underline{4b}$.

The reduction of \underline{lc} (X=OMe) ("base work-up") led directly (21% yield) to the 9-hydroxy-13-azasteroid $\underline{5}$, mp. \pm 180° decomp.; IR (KBr) 3280(OH), 1655

c) δ(CDCl₃) H
-C-N<
OR

(C=0)cm⁻¹; NMR (CDCl₃) & 3.0-3.4 (double t, \underline{H}_{14}) 3.5-3.9 (1H m, \underline{H}_{12}) 3.76 (s, \underline{CH}_3 0) 4.0-4.25 (1H m, \underline{H}_{12}) 6.55-6.85 (\underline{H}_2 and \underline{H}_4) 7.35-7.5(\underline{H}_1); thus illustrating the high nucleophilicity of the $\Delta 8$,9 bond in this system. Mild treatment of $\underline{5}$ with acid gave a 1:1 mixture of $\underline{4a}$ and $\underline{4b}$. On the contrary, in the reduction of \underline{Ic} ($\underline{F}=\overline{b}$) no trace of a \underline{g} -hydroxy-compound could be detected. Cyclication of a mixture of $\underline{3e}$ ($\underline{X}=\underline{OM}_2$) and $\underline{2e}$ ($\underline{X}=\underline{OM}_2$) gave the known left \underline{IS} -aza-equilenin methyl ether in \underline{IS} yield, mp. \underline{IS} - $\underline{I$

Acid treatment of compounds 2a, 3a, 2e (X=H) and 3e (X=H) afforded mainly the corresponding cyclization products as estimated via GLC-analysis and IR and MMR-spectroscopical data. In these reactions, however, considerable amounts of side-products were formed presumably as a consequence of the instability of the intermediate cyclic imminium form 6 and the relative inertness of the unactivated aromatic ring towards electrophilic substitution under the reaction circumstances⁵.

To examine the applicability of the imide reductive cyclization method in a different field, experiments were carried out in the tryptamine series. Tryptamine is known to condense with 2-pyrrolinones to the 11H-indolo-[3,2-g]-indolizine derivative $\underline{7}$ (n=1)⁶. When 5-ethoxy-2-pyrrolidinone⁷ was reacted with tryptamine in acetic acid the 11H-indolo-[3,2-g]-indolizine $\underline{7}$ (n=1) was obtained in 67% yield. The corresponding 6-ethoxy-2-piperidone⁷ afforded in a similar reaction 12H-indolo-[2,3-a]-quinolizine $\underline{7}$ (n=2)⁸ in 81% yield.

$$H_{3}CO$$
 $A = A = 9, 11$
 $A = A = 1, 11$
 $A = 1, 11$

The latter reaction most probably proceeds via 3,4-dihydro-2(1H)-pyridinone : 7 (n=2) being also obtained in comparable yield from the condensation of the This method of preparing 7 (n=2) circumvents the enamide with tryptamine. use of the difficultly prepared α -ketoadipic acid⁹.

Finally the reductive cyclization was applied to the homophthalimide 108. The corresponding iso-carbostyril derivative smoothly cyclized to 9 by refluxing in HCl-methanol in 86% yield; mp. 11 250-260° (dec.).

Various other ring closure reactions of compounds containing the structural elements of the types --COH-NR-CO- and --COEt-NR-CO- in the synthesis of alkaloidal systems are being currently explored.

REFERENCES:

- 1) a. T. Kametani, S. Tabaro and E. Karibe, Yakugaku Zasshi 83 1039 (1963), cf. CA.60 132236 (1964);

 - b. E. Wenkert, S. Garratt and K.E. Dave, Can.J.Chem. 42 489 (1964);
 c. G.C. Morrison, W. Cetenko and J. Shavel Jr., J.Org.Chem. 29 2771 (1964)
 d. W.E. Schleigh, A. Catala and F.D. Popp, J.Heterocycl.Chem. 2 379 (1965)
 e. S.V. Kessar, M. Singh and A. Kumar, Tetrahedron Letters (1965) 3245;
 f. A.J. Birch and G.S.R. Subba Rao, J.Chem.Soc.(C) 1965 3007;
 g. T. Kametani, R. Yanase and S. Takano, Yakugatu Kenkyu 37 23 (1966),
 c.f. CA.65 153206 (1966);
 h. J.C. Hubert W.N. Speckers and H.O. Huderen, Matabadae, Letters (1966).
 - h. J.C. Hubert, W.N. Speckamp and H.O. Huisman, Tetrahedron Letters (1969) 1553.
- 2) a. F.V. Brutcher Jr., W.D. Vanderwerff and B. Dreikorn, J.Org.Chem. 37 297-302 (1972);

 - b. Atta-Ur-Rahman, J.C.S. Perkin I 1972 731-5, 736-8; c. T. Kametani, K. Fukumoto, T. Terui, K. Yamaki and E. Taeuchi, J.Chem.Soc.(C) 1971 2709-11;
 - d. G.C. Morrison, W.A. Cetenko and J. Shavel Jr., J.Org.Chem. 36 3624-7 (1971);
- 3) J.C. Hubert, W. Steege, W.N. Speckamp and H.O. Huisman, Synth. Comm. 1 103 (1971).
- 4) M. Winn and H.E. Zaugg, <u>J.Org.Chem</u>. 33 3779 (1968).
- 5) The corresponding phtalimide derivatives of <u>2a</u> and <u>2e</u> (X=H) give nearly quantitative yields of cyclization product (as will be published elsewhere see also ref.3); in that case the intermediate is presumably a relatively stable immonium ion.
- 6) V. Bocchi, G. Casnati and G.P. Gardini, Tetrahedron Letters (1971) 683.
- 7) The synthesis of this compound will be published in our full paper.
- 8) S. Corsano and S. Algieri, Ann, Chim. (Roma) 50 75 (1960).
- 9) a. G. Hahn and A. Hansel, <u>Chem.Ber</u>. 71 2163 (1938); b. W. Mayer, R. Bachmann and F. Kraus, <u>Chem.Ber</u>. 88 316 (1955).
- 10) G.R. Clemo and G.A. Swan, J.Chem.Soc. (1946) 617.
- 11) C. Ribbens, Scient.Comm.Research Dept. N.V. Koninkl.Pharm.Fabrieken v/h Brocades-Stheeman & Pharmacia 10 9 (1960-1), cf. CA.56 7379 g (1962).